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(12.00 x 9.93) + (13.00 x 1.07) 100 Both isotopes contain 6 protons but different numbers of neutrons. Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 protons and 7 neutrons. CHEMISTRY ANSWER KEY 27 Na Is22s22p63s1 Mg 1s22s22p63s2 Al Is22s22p63s23pl Si Is22s22p63s23pl

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