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Gizmo. Practice balancing
chemical equations by changing
the coefficients of reactants and
products. As the equation is
manipulated, the amount of each
element is shown as individual

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atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Answer Key Chemical Equations

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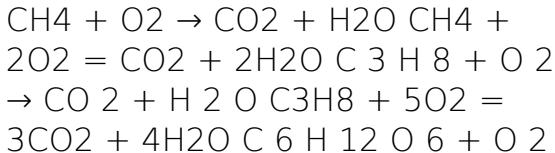
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look at the floating molecules
below the initial reaction: $H_2 + O_2$
How many atoms are
in a hydrogen molecule (H_2)?
 2 How many atoms are in an
oxygen molecule (O_2)? 2

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Student Exploration: Chemical
Equations Combustion - a fuel
reacts with oxygen to release
carbon dioxide, water, and heat.



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replacement - an element
replaces another element in a
compound. $\text{KCl} + \text{F}_2 \rightarrow \text{KF} + \text{Cl}_2$
 $2\text{KCl} + \text{F}_2 = 2\text{KF} + \text{Cl}_2$ $\text{Mg} + \text{HCl}$
 $\rightarrow \text{MgCl}_2 + \text{H}_2$, $\text{Mg} + 2\text{HCl} =$
 $\text{MgCl}_2 + \text{H}_2$ $\text{Cu} + \text{AgNO}_3 \rightarrow$

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chemical equation. In the
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Gizmo, look at the floating
molecules below the initial
reaction: $H_2 + O_2 \rightarrow H_2O$. 1. How
many atoms are in a hydrogen
molecule (H_2)? ___ 2. How many
atoms are in an oxygen molecule

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2018 5. A chemical equation is balanced if the number of each type of atom on the left side is equal to the number of each type

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on the right side. Is this reaction balanced? _____ Activity A:
Balancing equations Get the Gizmo ready: Check that the Synthesis reaction is selected and that all coefficients are set to one. (The coefficients are the numbers in the boxes.)

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Gizmo, look at the floating molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? ___ 2. How many atoms are in an oxygen molecule (O_2)?

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box and "H₂O" in the Products
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