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When the volume expands to 24.7 L, a peg stops the piston. Calculate  $\Delta U$  (the change in energy),  $q$  (the heat transferred to the system), and  $w$  (the work done by the system) when the expansion is isothermal and reversible. Answer:  $q = 0$  (isolated system)  $w = 0$  (no external pressure)  $\Delta U = 0$  (all isothermal expansions) 16.

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